

Update on Microbiology Research at the Waste Isolation Pilot Plant (WIPP)

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April 16, 2013

LA-UR-13-22647



BACKGROUND



CONCEPTUAL MODEL of MICROBIAL EFFECTS on NUCLEAR WASTE REPOSITORIES

- Microbial activity may affect actinide oxidation state
- Organic waste components, including high molecular weight organics (cellulose, plastic, and rubber), will be degraded
- The degradation of organics leads to production of
 - gas
 - organic ligands
- Microorganisms can adsorb actinides and serve as vectors for transport away from repository
- Microbial activity may affect backfill materials



ASSUMPTION VS. REALITY

ASSUMPTIONS VERSUS REALITY in SALT-BASED WASTE REPOSITORIES

ASSUMPTION	REALITY
Near-field homogeneity means microorganisms have access to all waste components	Near-field will be heterogeneous with microenvironments
Basic requirements for optimal growth (water, nutrients, substrates, electron donors and acceptors) are present and accessible	Basic requirements are not always met and rarely ideal for halophiles; heterogeneity and solubility limit accessibility
Cellulose will be completely degraded	Partial degradation may occur if cellulolytic organisms are present, active, and have access to cellulose

ASSUMPTIONS VERSUS REALITY in SALT-BASED WASTE REPOSITORIES

ASSUMPTION	REALITY
Gas generation from consumption of organics	Generation rates are likely optimistic; CO ₂ accounted for by MgO at WIPP
Hydrocarbon degradation will lead to gas generation	Known haloarchaeal hydrocarbon degraders exist, but all are obligate aerobes; unlikely survival of halophilic bacterial degraders in repository brine, but more research needed
All organisms adsorb actinides; all organisms are mobile	Differential sorption behavior; presence of EDTA diminishes sorption; some cells motile, some sessile; cells may lyse between near- and far-fields
Is far-field taken into consideration?	Far-field microbial activity far greater than near-field

MICROBIAL ECOLOGY of the WIPP



SURVIVAL of MICROORGANISMS at HIGH IONIC STRENGTH

- Must be able to maintain osmotic balance with external environment
 - Haloarchaea and some Bacteria (order *Halanaerobiales*) maintain high concentrations of K^+ or Cl^- internally
 - All other Bacteria and all eukaryotes use mostly organic solutes
- Maintaining osmotic balance is energetically costly
- All metabolic processes are limited by thermodynamics; weigh cost of maintaining osmotic balance with benefit of energy obtained through given metabolic reaction

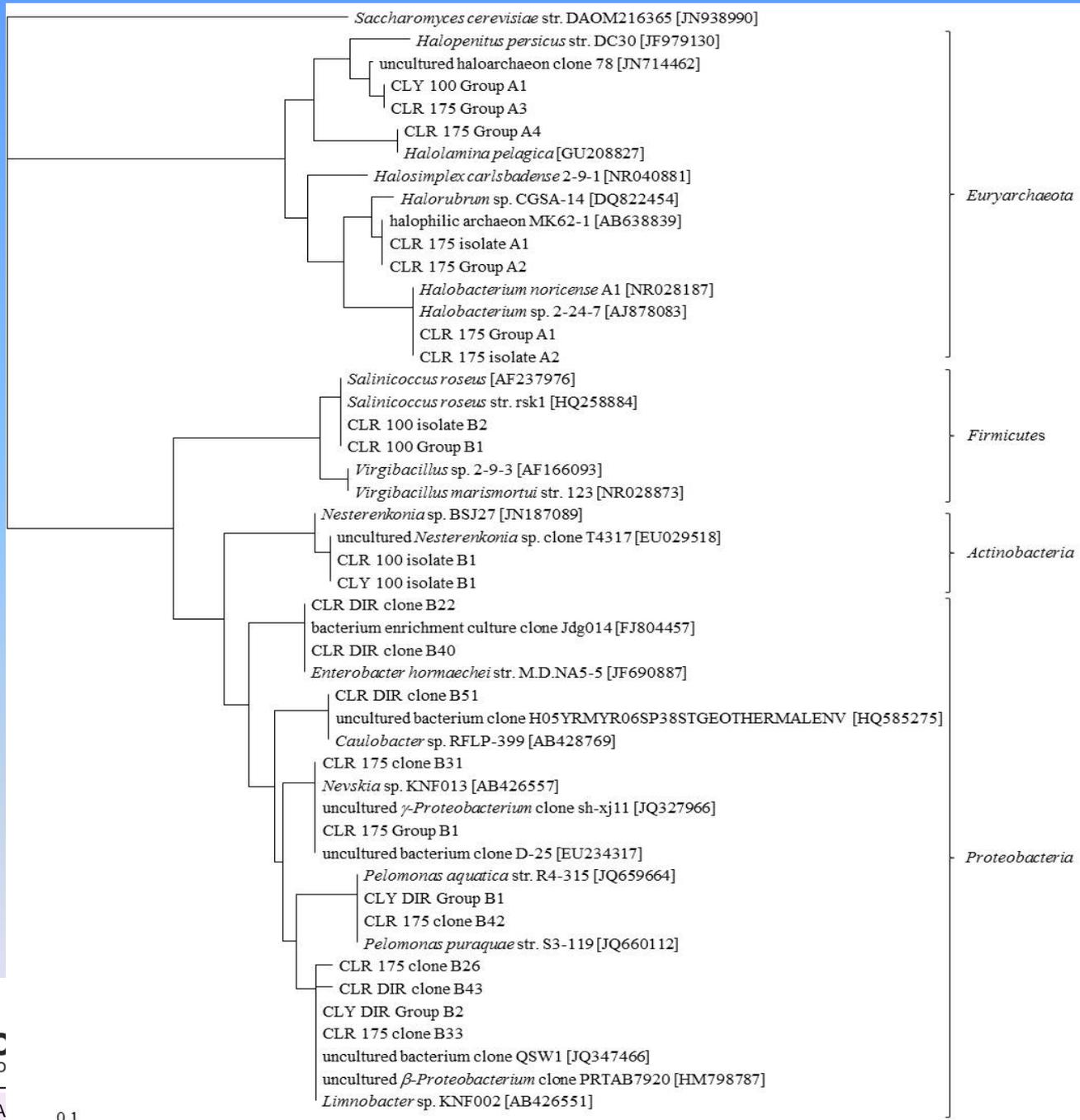
METABOLIC PROCESSES of HALOPHILIC MICROORGANISMS

MODE OF METABOLISM	HALOPHILIC BACTERIA	HALOARCHAEA	CONSTRAINTS OTHER THAN [SALT]
Aerobic	+	+	Oxygen availability
Nitrate Reduction	+	+	Inventory
Metal Reduction	+	-	Solubility
Sulfate Reduction	+	-	Few limits
Methanogenesis	-	-	Adequate substrate
Fermentation	+	+	Primary lytic species; substrate

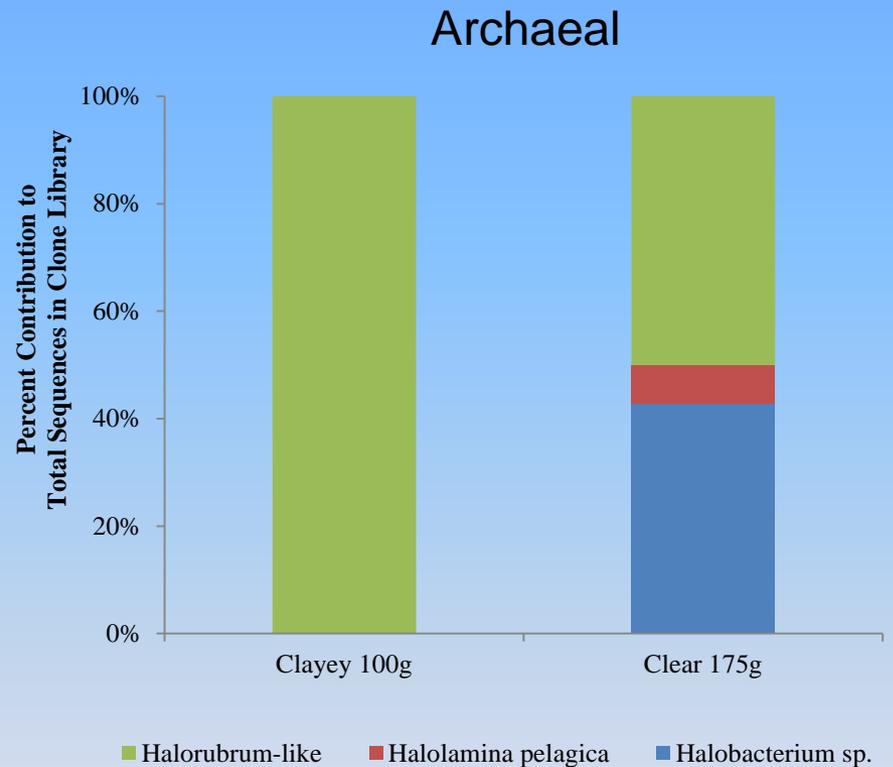
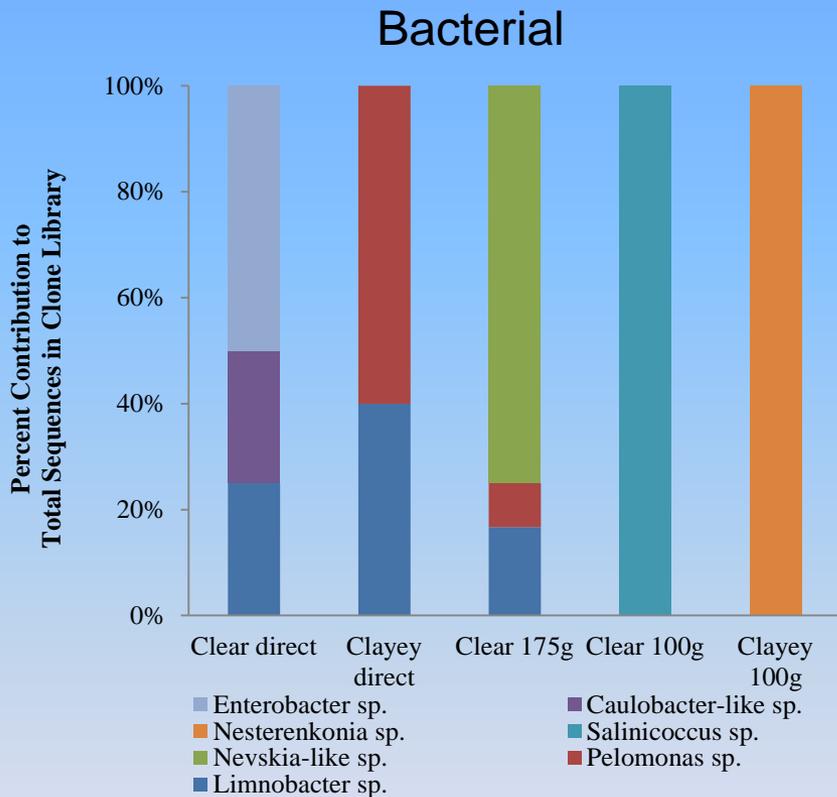
WIPP HALITE—”Near-field Environment”

- Extremely halophilic *Archaea* (all members of class *Halobacteria*):
 - *Halobacterium* (sorption experiments)
 - *Halorubrum*-like (unclassified)
 - *Halolamina*
 - *Natronomonas*
- Halotolerant to moderately halophilic *Bacteria*:
 - *Proteobacteria*
 - *Actinobacteria*
 - *Firmicutes* (Vreeland et al., 2000)
- Halotolerant to moderately halophilic *Fungi* (all members of phylum *Ascomycota*):
 - *Cladosporium*
 - *Engyodontium*
 - *Phoma*

ALL FROM AEROBIC INCUBATIONS
NO GROWTH IN ANAEROBIC INCUBATIONS



HALITE PROKARYOTIC DIVERSITY



HALITE EUKARYOTIC DIVERSITY

- Three fungal isolates:
 - *Cladosporium*
 - *Engyodontium*
 - *Phoma*
- All obligate aerobes



COMPARISON of SALADO and GORLEBEN HALITES— BACTERIA

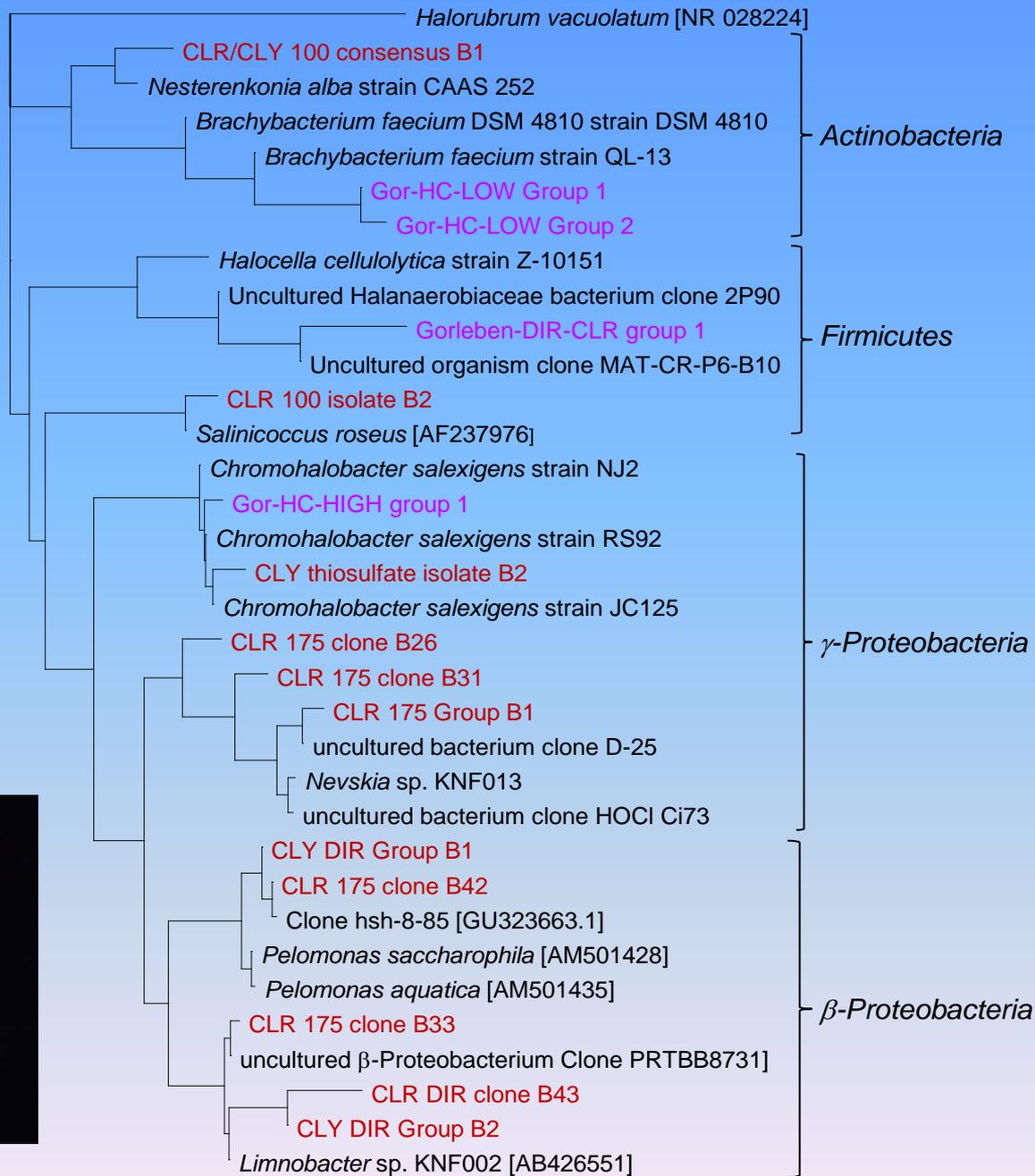
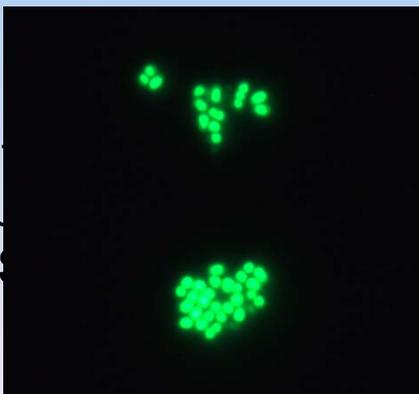
Shared organism:

Chromohalobacter

Not found in Salado:

Brachybacterium, *Halocella*-like

Archaeal results
negative for Eukaryotes
direct extract, Salado
in culture



0.1

Unclassified LA-UR 13-

WIPP GROUNDWATERS

- Far-field environment
- Range of ionic strengths (~1.5-5 M)
- Low in DOC, nitrate; high in sulfate (25-74 mM), Mg^{2+} , K^+ , Ca^{2+}
- *Bacteria* with diverse metabolic capability:
 - Nitrate reducers
 - Metal reducers
 - Sulfate reducers
 - **Fermenters** (including extremely halophilic *Bacteria*)
- When metals are present, incubations result in precipitation of metal-sulfide; both metal-reducers and sulfate-reducers can be detected

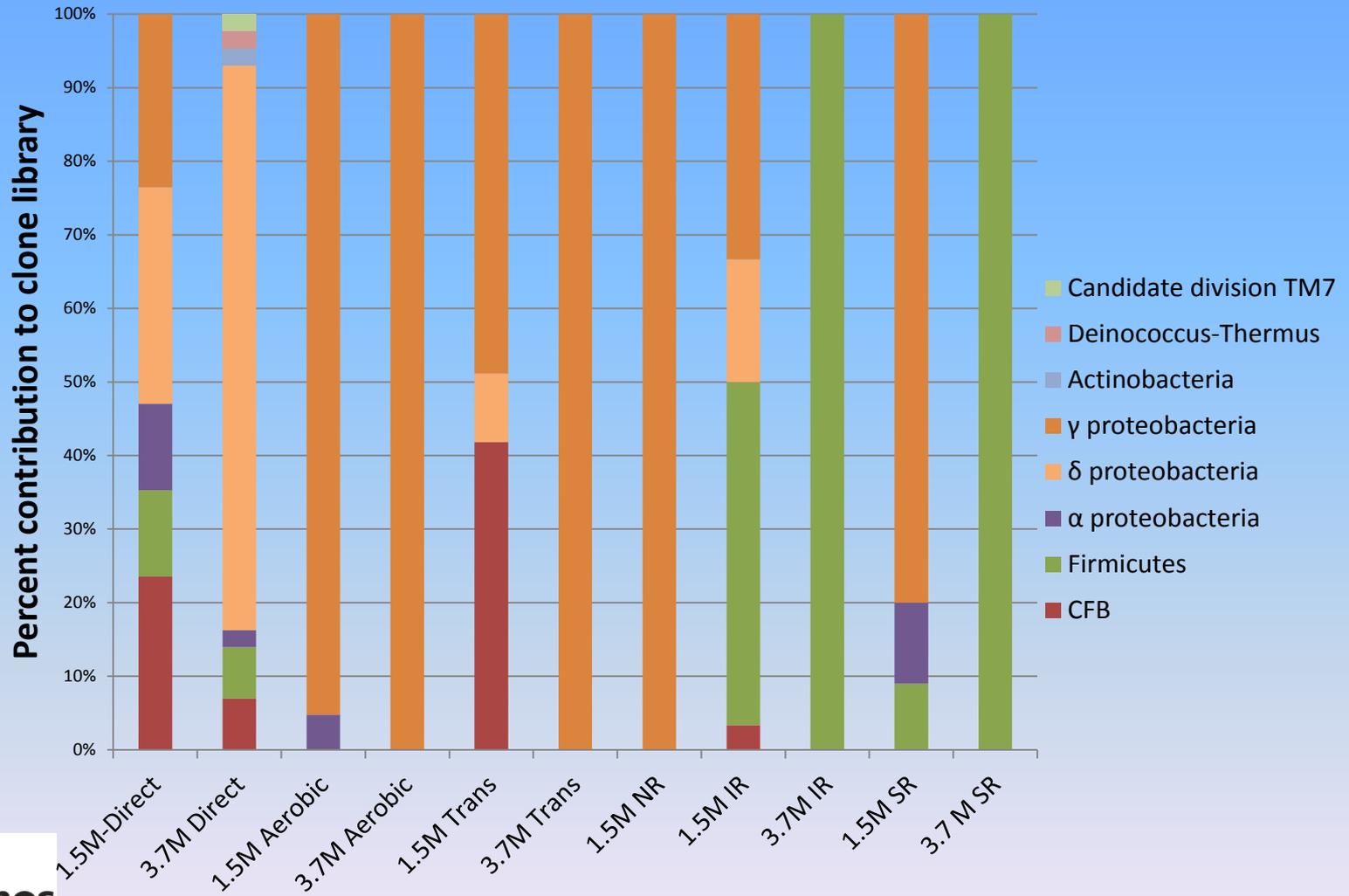
WIPP GROUNDWATERS—AEROBIC

- Moderate to extremely halophilic *Archaea* (all order *Halobacteriales*):
 - *Halococcus*
 - *Natrinema*
- Halotolerant to extremely halophilic *Bacteria*:
 - *Proteobacteria* (*Chromohalobacter* spp.)
 - *Firmicutes* (*Virgibacillus*, *Pontibacillus*, *Bacillus*)
- Halotolerant *Fungi* (only in 5 M groundwater; not yet characterized)

WIPP GROUNDWATERS—ANAEROBIC

- Moderate to extremely halophilic *Archaea* (all *Halobacteriales*), viable but not growing anaerobically:
 - *Natrinema*
 - *Haloferax*
 - *Haloarcula*
- Halotolerant to moderately halophilic *Bacteria*:
 - *Proteobacteria* (including denitrifiers, iron- and sulfate-reducers)
 - *Bacteroidetes*
 - *Firmicutes* (including iron-reducers and fermenters)

GROUNDWATER COMPARISON-BACTERIA

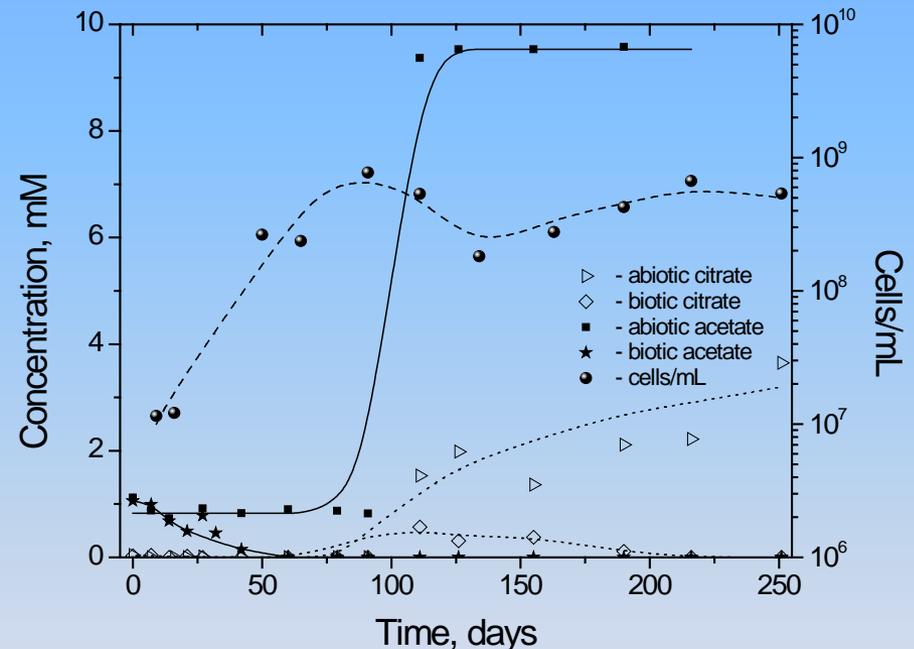


BIODEGRADATION of ORGANIC WIPP WASTE



DEGRADATION of LOW MOLECULAR WEIGHT ORGANICS—COMPLEXING AGENTS

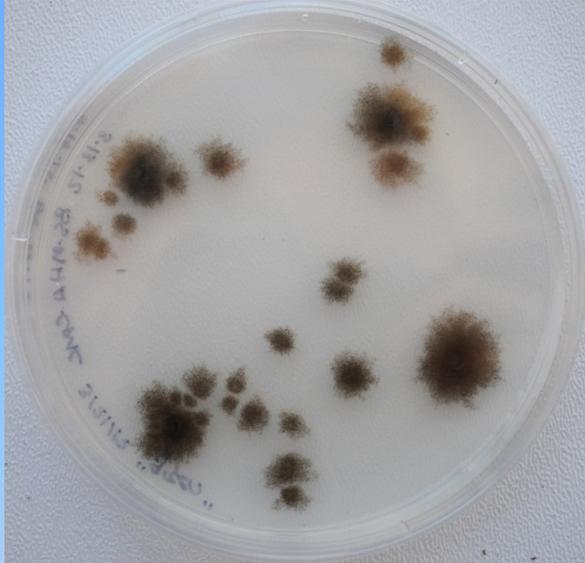
- Aerobic degradation of acetate, oxalate, and citrate
- Degradation of citrate under denitrifying conditions not shown (shown by Francis et al., 2000, using organism enriched from area brine lake); co-oxidation of citrate with acetate being tested



DEGRADATION of HIGH MOLECULAR WEIGHT ORGANICS—CELLULOSES

- Degradation of cellulose and exhibition of cellulase activity under aerobic conditions
 - two isolates (fungal and bacterial) currently growing on solid agar with either carboxymethylcellulose or Kimwipes as sole carbon source
 - cellulase-positive bacterial isolates (*Salinicoccus*, *Nesterenkonia*)
- Potential for anaerobic fermentation in far-field
 - glucose fermenter isolated (*Halanaerobium* sp.)
 - sequences related to *Halocella* detected

AEROBIC CELLULOSE UTILIZATION



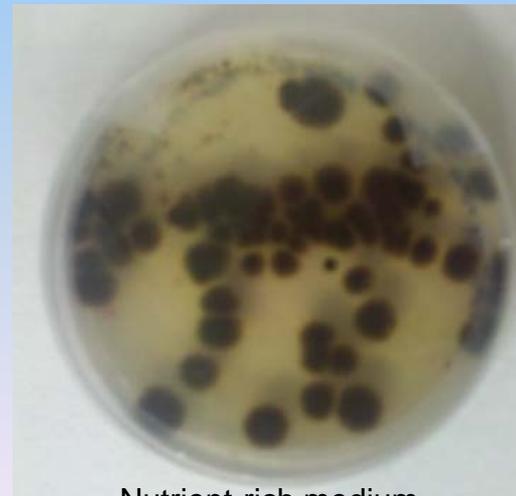
carboxymethylcellulose



No Kimwipe

Kimwipe

Cladosporium sp.
pH range: 5-10, optimum 6
NaCl range: 0-20%, optimum 10%



Nutrient-rich medium

METAL REDUCTION and PRECIPITATION in HIGH IONIC STRENGTH SYSTEMS

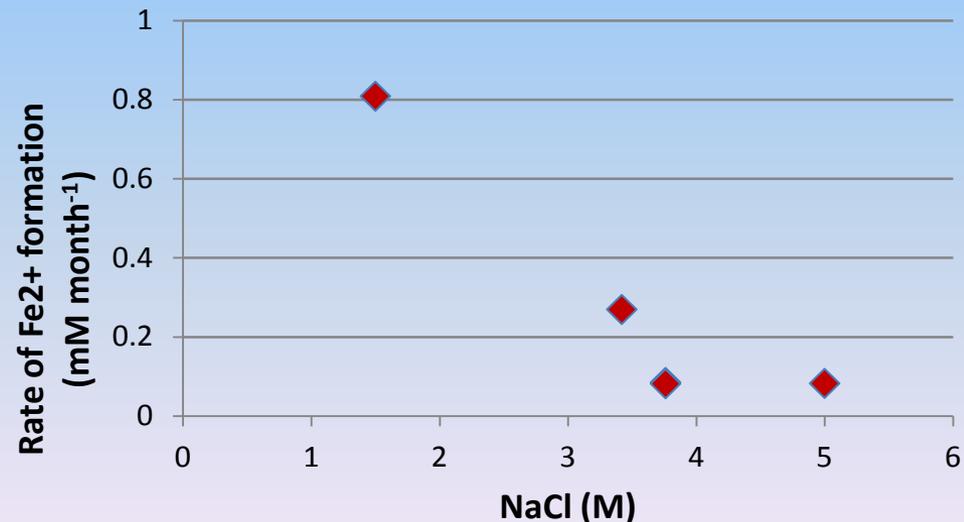


IMPLICATIONS for SALT-BASED WASTE REPOSITORIES

- Ability to reduce iron may translate to ability to reduce actinides
- Reduced iron precipitates with sulfide
- Reduced sulfate precipitates both oxidized and reduced iron species
- Metal-sulfide precipitate removes toxic sulfide
- Microorganisms induce metal precipitation as carbonate species, regardless of oxidation state

METAL REDUCTION in GROUNDWATER INCUBATIONS

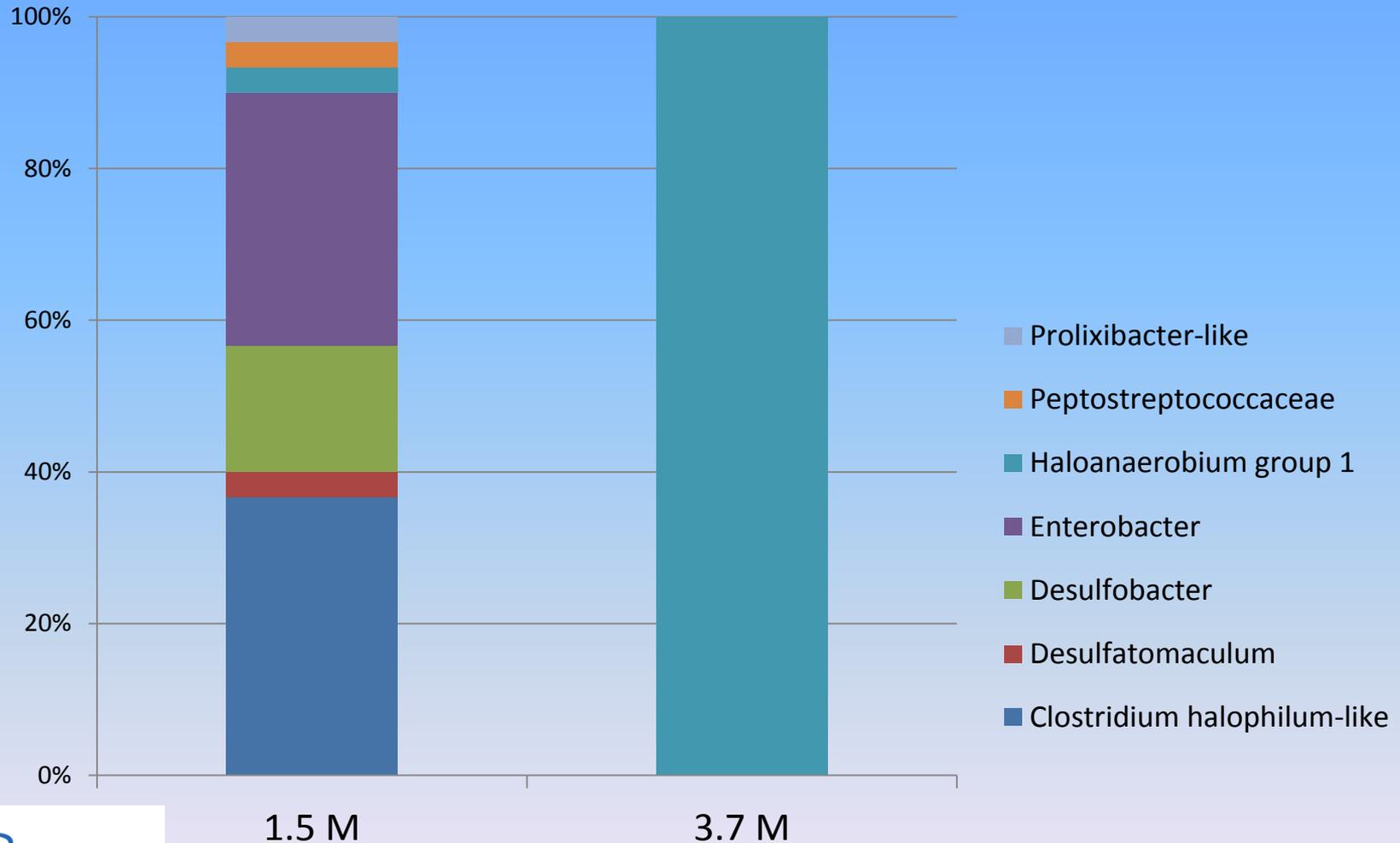
- Fe(III) solubility in saline systems is low; usually associated with POM and sediment in marine systems
- Iron reduction observed in *abiotic* incubations as function of ionic strength; slower rate than *microbially-induced* reduction
- Rates of reduction decreases with increasing [NaCl]



ORGANISMS FOUND in IRON-REDUCING INCUBATIONS

- 1.5 M: consortium of fermenters, iron-reducers, and sulfate-reducers (*Bacteroidetes*, *Halanaerobium*, *Clostridium*, *Peptostreptococcus*, *Desulfobacter*)
- 3.7 M: mixed culture of halophilic *Bacilli* (*Virgibacillus*, *Pontibacillus*); likely obligate aerobes
- 5 M: mixed culture of *Archaea* (*Haloferax*, *Haloarcula*) and *Bacteria* (see 3.7 M)

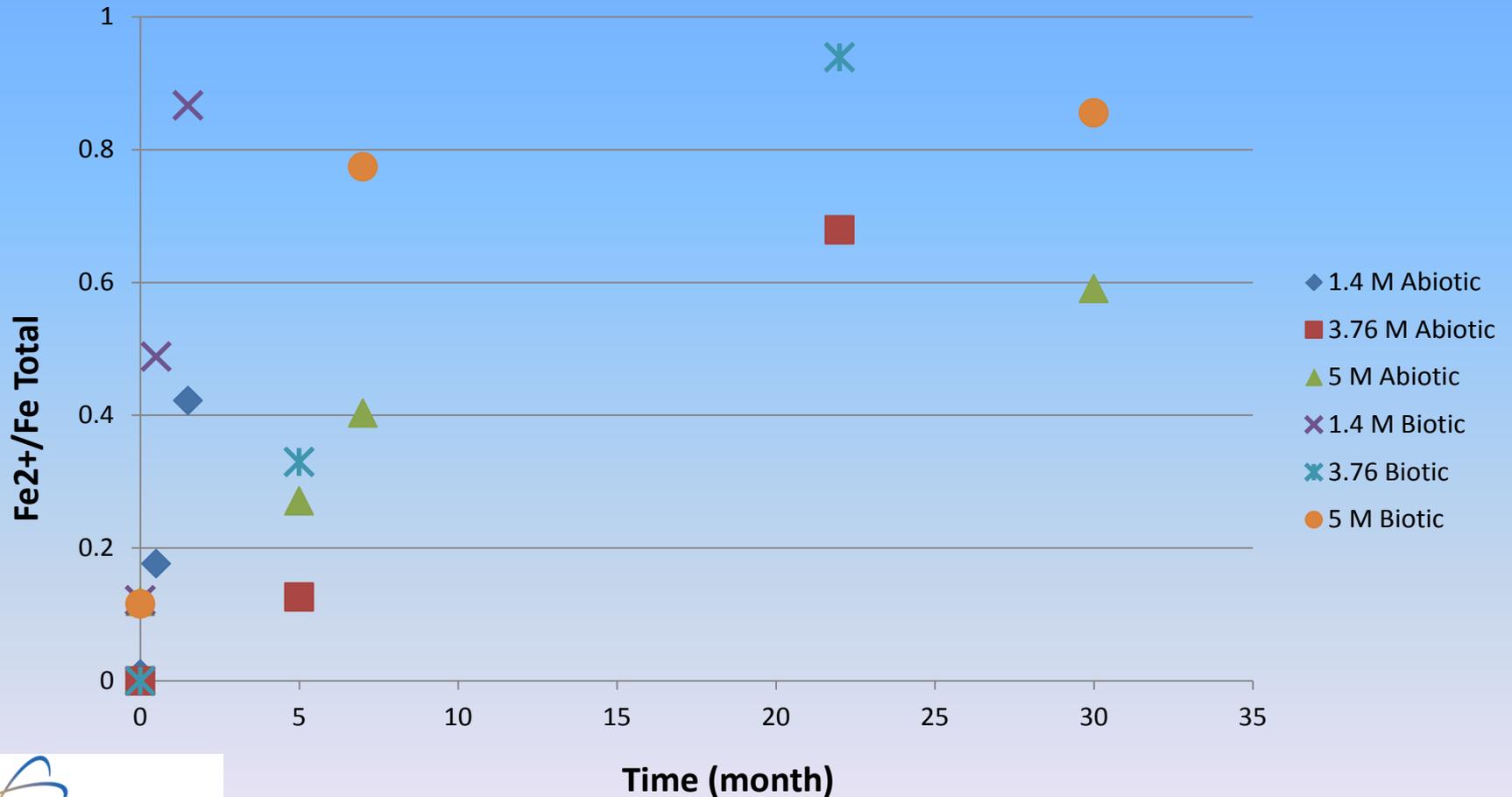
MICROBIAL DIVERSITY ASSOCIATED WITH IRON REDUCTION



POSSIBLE MODES of IRON REDUCTION at HIGH IONIC STRENGTH

- Direct reduction: appears limited to lower ionic strengths, both iron and sulfate reducers (1.5 M culture)
- Indirect reduction via:
 - Fermentation
 - Sulfate reduction
 - Release of reducing agents during sporulation or halocyst formation?
- Microbially-induced precipitation as iron carbonate, metal sulfide

IRON REDUCTION in BIOTIC and ABIOTIC INCUBATIONS at DIFFERENT IONIC STRENGTH

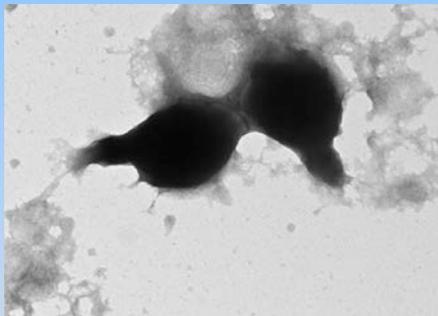
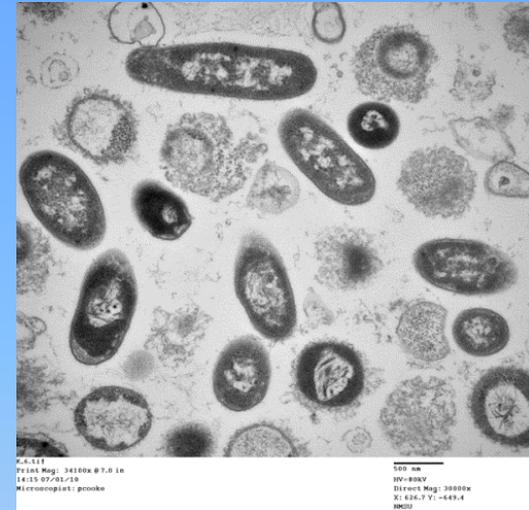


BIOSORPTION

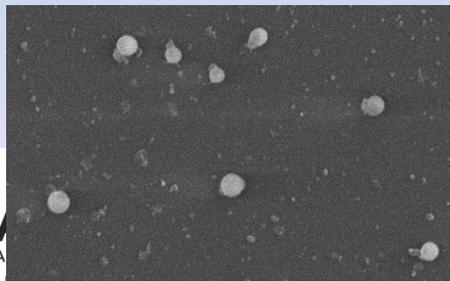


BIOSORPTION– TEST ORGANISMS

Chromohalobacter sp.—isolated from area groundwater, borehole seep, WIPP and Gorleben halites; strain used tolerates pH 5-9, 0.9-4.3 M NaCl; 0.3-0.5 x 1.5-2 μm size

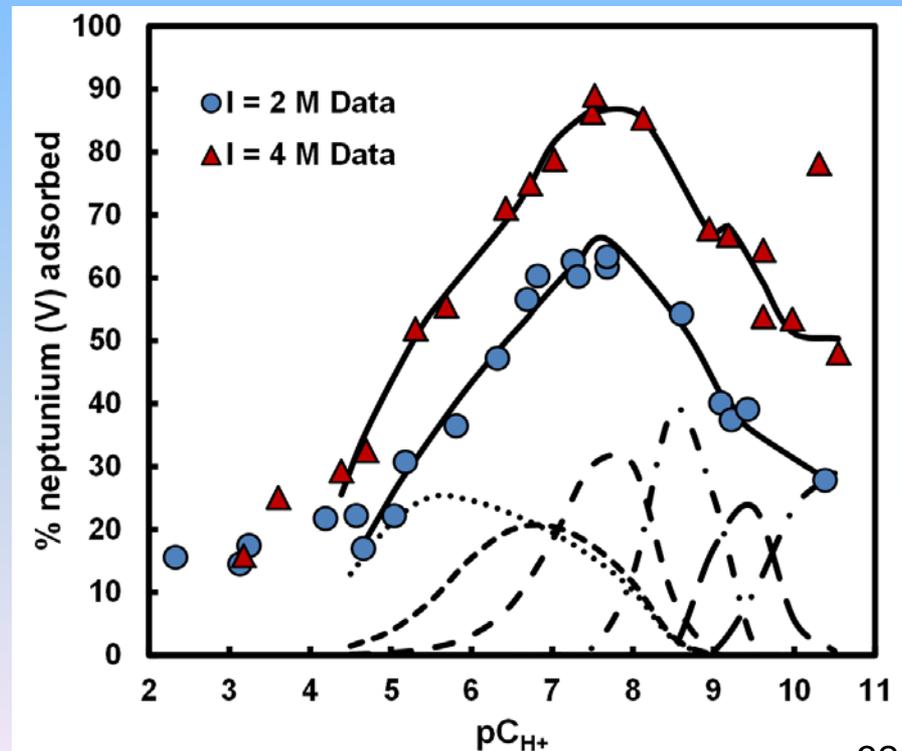


Halobacterium noricense—isolated from incubations of halite in generic media and in WIPP brines; detected in other subterranean salts worldwide (including Germany); requires 2.5-5 M NaCl, tolerates pH 6-10; 0.3-1.5 x 0.3-1.5 μm size

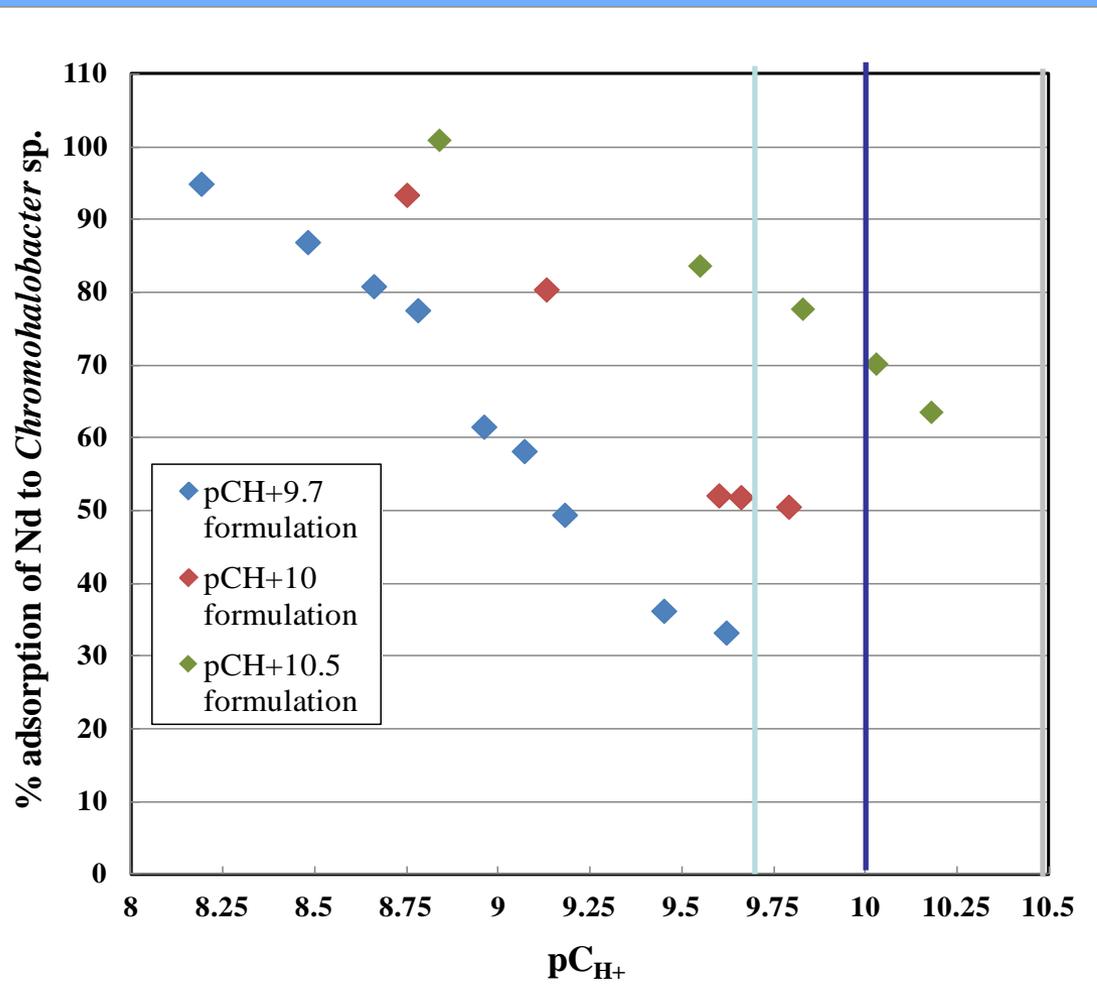


BIOSORPTION—Np onto *Chromohalobacter*

- Titration data at 2 and 4 M similar to each other
- Titration data similar to other organisms at low ionic strength
- Best-fit model invokes 4 surface complexation sites
 - Differences observed in sorption at 2 and 4M over pC_{H^+} range
 - Model invoking carbonate complexation reactions fits best at higher pC_{H^+}



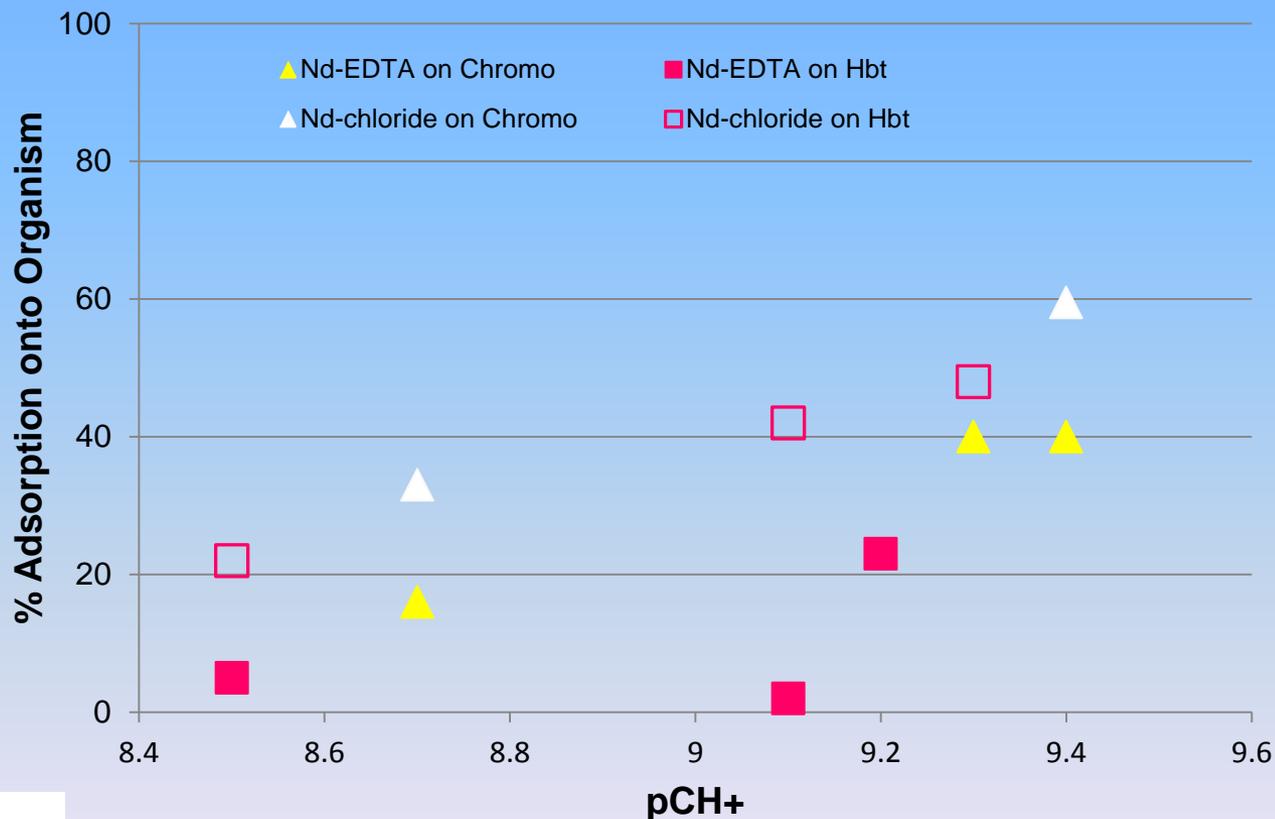
BIOSORPTION—Nd onto *Chromohalobacter*



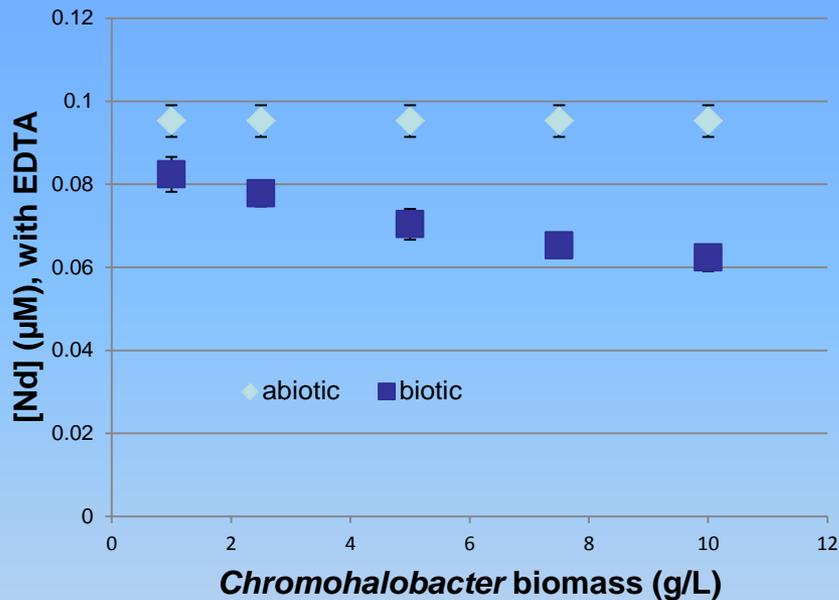
- Nd sorption to *Chromohalobacter* in 3 reacted ERDA formulations; carbonate in system
- Significant differences in sorption behavior as function of pC_{H^+} and formulation (Mg effect)
- Note: this is at limit of *Chromohalobacter* pH tolerance

ASSOCIATION of Nd with CELLS as a FUNCTION of pC_{H^+}

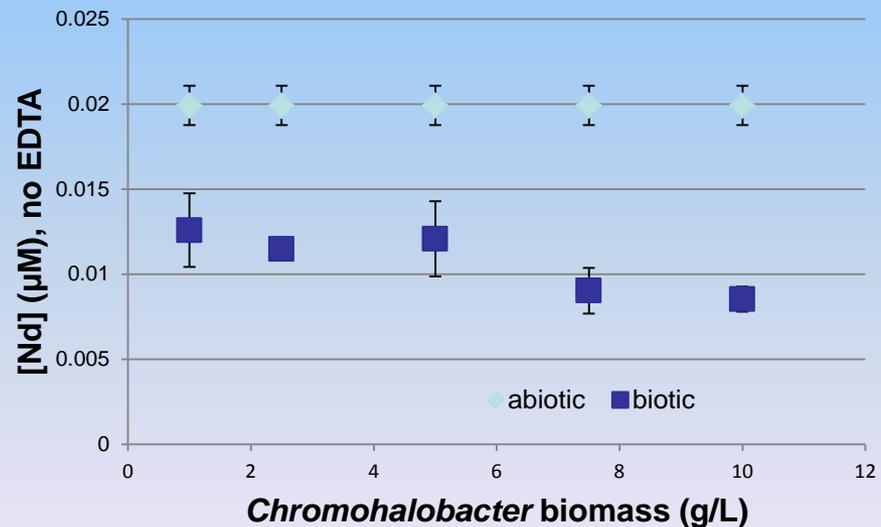
Complexation with EDTA decreases sorption



ASSOCIATION of Nd with CELLS as a FUNCTION of BIOMASS



Sorption increases with increasing biomass



SUMMARY of BIOSORPTION DATA at EXPECTED WIPP pC_{H^+}

- **Nd(III)**
 - *Archaea*: 25-50% sorption
 - *Bacteria*: 40-65% sorption
 - EDTA reduced sorption in both cases by ~ 40%
- **Th(IV)**
 - *Archaea*: 40-65% sorption
 - *Bacteria*: 81% (no EDTA) and 57% (EDTA) sorption
 - At $pC_{H^+} > 9$, cannot differentiate between sorption and precipitation, due to colloids and slow hydrolysis

CONCLUSIONS

- Structural and functional diversity of WIPP microbial communities differs between near- and far-field environments
- *Aerobic* degradation of organic complexing agents and cellulose by WIPP-indigenous organisms occurs; *Anaerobic* degradation of organic complexing agents has not been shown with halite-derived organisms; bacterium similar to only known halophilic, anaerobic, cellulolytic microorganism detected in WIPP groundwater and associated with Gorleben halite
- Both abiotic and microbially-mediated metal reduction occur in hypersaline systems; rates dependent upon [Na]
- Actinide sorption to microorganisms shows ionic strength, pC_{H^+} , organism, and actinide dependence

ACKNOWLEDGMENTS

This work is sponsored by the US Department of Energy—Carlsbad Field Office

We especially thank:

- Russ Patterson, DOE program manager and support
- WIPP and Washington TRU Solutions for sampling assistance

